Amendments to the Claims

The following listing of claims replaces all prior versions and listings of claims in this application:

Claims 1 to 18. (Cancelled)

19. (Currently Amended) A method for electroplating [[a]] metal deposits on [[a]] substrates which comprises contacting a plurality of such substrates with a solution which comprises:

water;

[[a]] metal ions in an amount sufficient to provide [[a]] the metal deposits on a platable such substrates;

a complexing agent of an organic compound having between 4 and 18 carbon atoms which compound includes at least two hydroxyl groups and a five or six membered ring that contains at least one oxygen atom, with the compound being present in an amount sufficient to complex the metal ions to render [[it]] them soluble in the solution and to inhibit oxidation of the metal ions; and

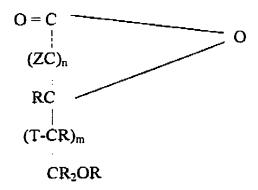
a pH of the solution in the range of between 3.5 and 5.5, adjusted, if necessary, by the addition of a suitable pH adjusting agent;

wherein the complexing agent and metal ions are present in a concentration ratio of between about 2:1 and 9:1 to reduce or minimize agglomeration of the substrates during the electroplating; and

passing a current though the solution to provide metal electrodeposits on the substrates without causing significant agglomeration of such substrates during the electroplating.

- 20. (Cancelled)
- 21. (Previously Presented) The method of claim 19, wherein the complexing agent has the structure:

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wherein each R is the same or different and is hydrogen or a lower alkyl group of 1 to 3 carbon atoms, T is R, OR, or O=P(OR)₂-, Z is O= or RO-, n is 2-4 and Z can be the same or different in each occurrence in the compound, and m is 1-3, or the complexing agent is a soluble salt of such structure.

- 22. (Previously Presented) The method of claim 19, wherein the complexing agent is ascorbic acid, isoascorbic acid, dehydroascorbic acid, glucoascorbic acid, galacturonic acid, glucoronic acid, or a salt thereof, or is derived from a ketogluconate or heptagluconate and is present in an amount of about 25 to 200 g/l.
- 23. (Currently Amended) The method of claim 19, wherein the metal ions [[is]] are tin ions and [[is]] are added to the solution as [[a]] stannous alkyl sulfonate salts, [[a]] stannous sulfate salts, [[a]] stannous chloride salts, [[a]] stannous ascorbate salts, or stannous oxides and [[is]] are present in an amount of between about 5 and 100 g/l.
- 24. (Currently Amended) The method of claim 23, which further comprises wherein the metal ions include a divalent lead salt in an amount sufficient to deposit a tin-lead alloy from the solution.
- 25. (Currently Amended) The method of claim 19, which further comprises wherein the solution includes a conductivity salt in an amount sufficient to increase the conductivity of the solution.

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- 26. (Previously Presented) The method of claim 25, wherein the conductivity salt is an alkali or alkaline metal sulfate, sulfonate, or acetate compound.
- 27. (Currently Amended) The method of claim 19, which further comprises wherein the solution includes a surfactant in an amount sufficient to enhance deposit quality and grain structure.
- 28. (Previously Presented) The method of claim 27, wherein the surfactant is an alkylene oxide condensation compound and is present in an amount of about 0.01 to 20 g/l.
- 29. (Currently Amended) The method of claim 19, further comprising wherein the solution includes an agent to promote anode dissolution.
- 30. (Previously Presented) The method of claim 29, wherein the agent to promote anode dissolution is as potassium methane sulfonate, ammonium chloride or a metal sulfide salt.
- 31. (Previously Presented) The method of claim 19, wherein the substrates are composite articles having electroplatable and non-electroplatable portions, the pH adjusting agent is an acid or a base and the pH is adjusted to the range of about 3.5 to 5.5 to enable electroplating of the electroplatable portions of the articles without deleteriously affecting the non-electroplatable portions.
- 32. (Currently Amended) In a method for electroplating [[a]] metal deposits on [[a]] substrates by contacting a plurality of such substrates with a solution, the improvement which comprises formulating a solution which comprises:

water;

[[a]] metal ions in an amount sufficient to provide [[a]] the metal deposits on a platable such substrates;

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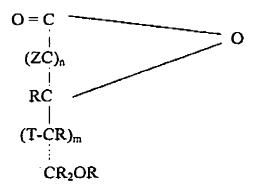
a complexing agent of an organic compound having between 4 and 18 carbon atoms which compound includes at least two hydroxyl groups and a five or six membered ring that contains at least one oxygen atom, with the compound being present in an amount sufficient to complex the metal ions to render [[it]] them soluble in the solution and to inhibit oxidation of the metal ions; and

a pH of the solution in the range of between 3.5 and 5.5, adjusted, if necessary, by the addition of a suitable pH adjusting agent;

wherein the complexing agent and metal ions are present in a concentration ratio of between about 2:1 and 9:1 to reduce or minimize agglomeration of the substrates during the electroplating;

so that significant agglomeration of such substrates is avoided when a current is passed though the solution to provide the metal electrodeposits on the substrates.

- 33. (Previously Presented) The method of claim 32, wherein the metal ions are tin ions and are present in an amount of between about 5 and 100 g/l.
- 34. (Previously Presented) The method of claim 32, wherein the complexing agent is present in an amount of about 25 to 200 g/l and has the structure:



wherein each R is the same or different and is hydrogen or a lower alkyl group of 1 to 3 carbon atoms, T is R, OR, or O=P(OR)₂-, Z is O= or RO-, n is 2-4 and Z can be the same or different in

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each occurrence in the structure, and m is 1-3, or the complexing agent is a soluble salt of such structure.

- 35. (Previously Presented) The method of claim 32, wherein the complexing agent is ascorbic acid, isoascorbic acid, dehydoascorbic acid, glucoascorbic acid, galacturonic acid, glucoronic acid, glucose-6-phosphate, or a salt thereof, or is derived from a ketogluconate or heptagluconate and is present in an amount of about 25 to 200 g/l.
- 36. (Currently Amended) The method of claim 32, further comprising wherein the solution includes one of a conductivity salt of an alkali or alkaline metal sulfate, sulfonate, or acetate compound, or a surfactant of an alkylene oxide condensation compound in an amount of about 0.01 to 20 g/l, or both.
- 37. (Previously Presented) The method of claim 32, wherein the substrates are composite articles having electroplatable and non-electroplatable portions, the pH adjusting agent is an acid or a base and the pH is adjusted to the range of about 3.5 to 5.5 to enable electroplating of the electroplatable portions of the articles without deleteriously affecting the non-electroplatable portions.

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